

Course Number and Name													
BME302 – THERMODYNAMICS													
Credits and Contact Hours													
4&60													
Course Coordinator's Name													
Mr.S.Manavalan													
Text Books and References													
TEXT BOOKS:													
1. P.K.Nag-Basic and Applied Thermodynamics-Tata McGraw Hill Publishing Company, 2002													
2. R.K.Rajput-Engineering Thermodynamics-Laxmi Publications													
REFERENCES:													
1. S.C.Somasundaram-Thermal Engineering-New Age International (P) Ltd,1996													
2. Y.V.C.Rao-An Introduction to Thermodynamics-New Age International (P) Ltd, 2004													
3. Yunus A.Cengel-Thermodynamics-International Edition, 2006													
4. bookboon.com/en/engineering-thermodynamics-ebook													
Course Description													
To achieve an understanding of principles of thermodynamics and to be able to use it in accounting for the bulk behavior of the simple physical systems.													
Prerequisites							Co-requisites						
MATHEMATICS –I &II							Nil						
required, elective, or selected elective (as per Table 5-1)													
Required													
Course Outcomes (COs)													
CO1	Solve first law thermodynamics based types of problems.												
CO2	Solve second law thermodynamics based types of problems.												
CO3	Understand Thermodynamic properties of pure substances												
CO4	understand Thermodynamic relations & gas laws												
CO5	Extend the ideas in implementation of mini/major project												
CO6	Understand combustion of fuels												
Student Outcomes (SOs) from Criterion 3 covered by this Course													
	COs/SOs	a	b	c	d	e	f	g	h	i	j	k	l
	CO1	H	H	H			H		L	H		M	M
	CO2						H		L			M	
	CO3						H		L			M	M
	CO4						H		L			M	M
	CO5						H		L			M	M

	CO6						H		L			M	
List of Topics Covered													
UNIT-I	BASIC CONCEPTS AND FIRST LAW OF THERMODYNAMICS												12
<p>Concept of continuum, Thermodynamic systems-closed, open and control volume, Thermodynamic properties, path, point functions, process - Quasistatic processes, cycle, work, modes of work, heat, temperature, Zeroth law of thermodynamics, First law of Thermodynamics-applications to open and closed systems, internal energy, Specific heats C_p, C_v, enthalpy, steady and unsteady flow conditions.</p>													
UNIT-II	SECOND LAW OF THERMODYNAMICS												12
<p>Kelvin's and Clausius statements, Reversibility, Applications - Carnot cycle, Reversed Carnot cycle, heat engines, Refrigerators, heat pumps, Concept of Entropy, Clausius Inequality, Principle of increase of entropy, Carnot theorem, Entropy and irreversibility, Available energy, Availability, Gibbs and Helmholtz functions</p>													
UNIT III	THERMODYNAMIC PROPERTIES OF PURE SUBSTANCES												12
<p>Thermodynamic Properties Of Pure Substances in solid, liquid and vapour phases, P-V, P-T, T-V, T-S, H-S diagrams, PVT surfaces, steam table of thermodynamic properties, Calculations of properties, Work done and heat transferred in non flow and flow processes.</p>													
UNIT IV	THERMODYNAMIC RELATIONS & GAS LAWS												12
<p>Exact differential, Tds relations, Maxwell, Clausius-Clapeyron equation, Joule Thomson Coefficient, Avagadro's Law, Vanderwaal's equation of state, mole concept, molar volume, equivalent weight, properties of mixture, Dalton's law of partial pressure, Amagat law, Enthalpy and specific heat, Molecular weight of gas mixture.</p>													
UNIT V	COMBUSTION OF FUELS												12
<p>Heating value of fuels, Combustion equations, Theoretical and excess air, Air-fuel ratio, Exhaust gas analysis, adiabatic flame temperature.</p>													